



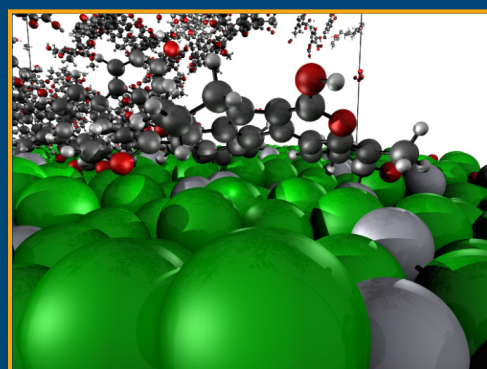
Scientific Computing & Modelling

Quality Software.  
Quantum Science.

# ADF Modeling Suite for Petrochemistry

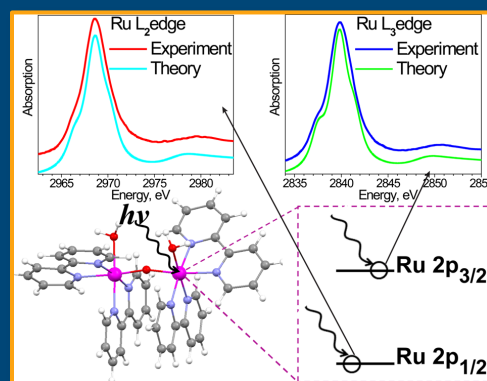
## Easy integrated modeling solution

- Atomistic (DFT, DFTB, MOPAC, ReaxFF) & continuum (COSMO-RS)
- Cross-platform (Windows, Linux, Mac)
- Single graphical interface
- Parallel install out of the box
- Expert staff and support



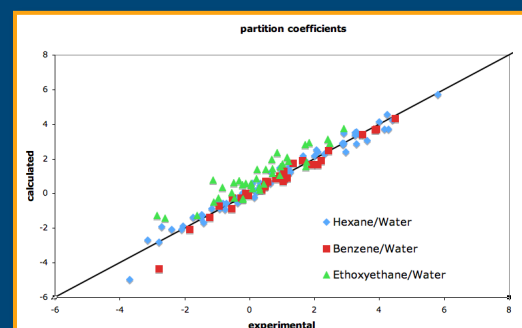
## Catalysis

- Surfaces 2D: no dipole issues, solvation
- Molecular + Periodic DFT
- Chemical bond analysis: rational design
- Spectroscopy: XAS, EPR, NMR, UV/VIS
- Strong in transition metal chemistry
- All-electron basis, relativistic effects
- Semi-empirical: MOPAC, DFTB
- Reactive MD: chemistry of large, complex systems



## COSMO-RS: fluid thermodynamics

- Quantum-based predictions
- VLE, LLE, logP, solubilities
- COSMO-SAC-2013: improved VLE
- Compounds database

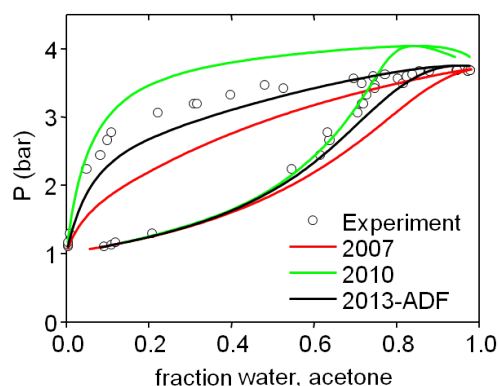




# Petrochemistry Modeling Research Highlights

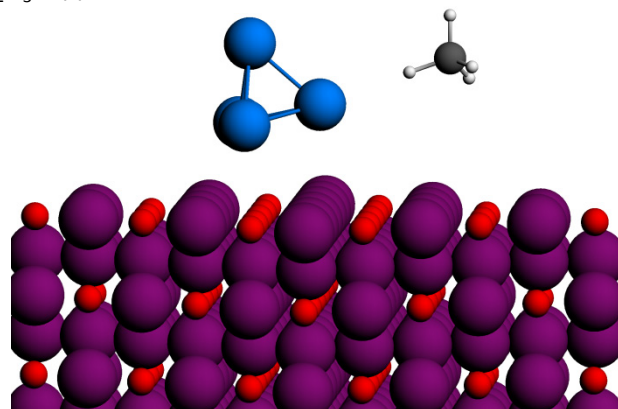
## Scientific Computing & Modelling

Improved thermodynamic properties with COSMO-SAC-2013



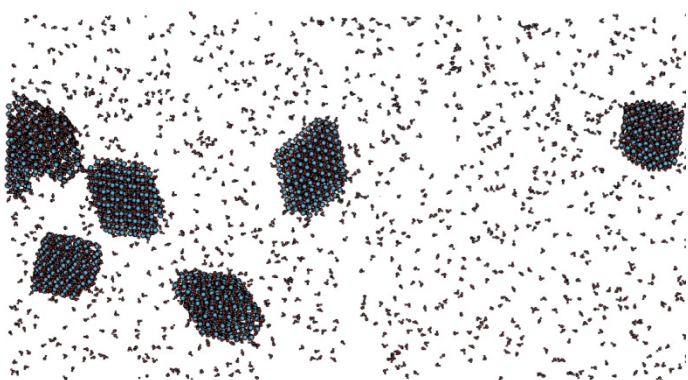
R. Xiong, S. I. Sandler, and R. I. Burnett, *An Improvement to COSMO-SAC for Predicting Thermodynamic Properties* [Ind. Eng. Chem. Res.](#) **53**, 8265 (2014)

$\text{Al}_2\text{O}_3$  support acts as e-donor for methane activation over Ni



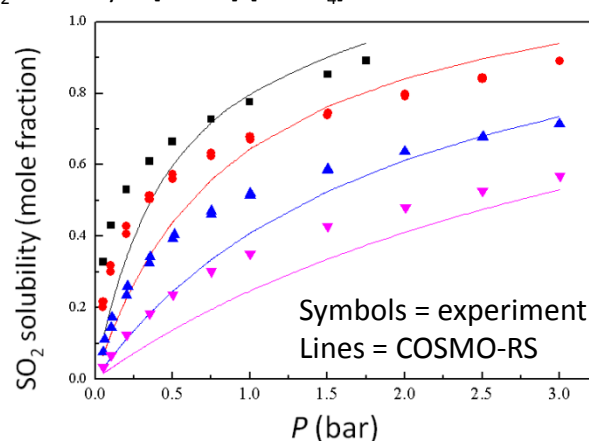
J. Li, E. Croiset, and L. Ricardez-Sandoval, *Effect of Metal-Support Interface During CH<sub>4</sub> and H<sub>2</sub> Dissociation on Ni/ $\gamma$ -Al<sub>2</sub>O<sub>3</sub>: A DFT Study*, [J. Phys. Chem. C](#) **117**, 16907 (2013)

Crystal growth: oriented attachment of TiO<sub>2</sub> nanoparticles



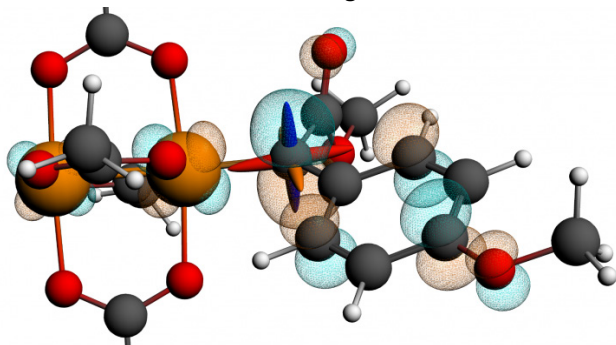
M. Raju, A. C. T. van Duin, and K. A. Fichtorn, *Mechanisms of Oriented Attachment of TiO<sub>2</sub> Nanocrystals in Vacuum and Humid Environments: Reactive Molecular Dynamics* [Nano Lett.](#) **14**, 1836 (2014)

SO<sub>2</sub> solubility in [BMIM]<sup>+</sup>[MeSO<sub>4</sub>]<sup>-</sup> with COSMO-RS



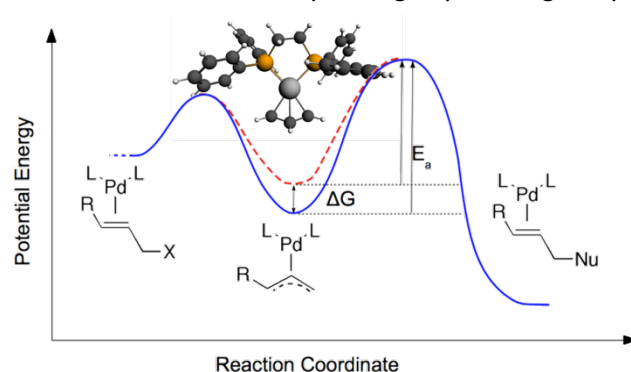
Z. Lei, C. Dai, and B. Chen, *Gas Solubility in Ionic Liquids* [Chem. Rev.](#), **114**, 1289 (2014)

Relativistic DFT locates <sup>13</sup>C NMR signal of dirhodium carbene



K. P. Kornecki, J. F. Briones, V. Boyarskikh, F. Fullilove, J. Autschbach, K. E. Schrote, K. M. Lancaster, H. M. L. Davies, and J. F. Berry, *Direct Spectroscopic Characterization of a Transitory Dirhodium Donor-Acceptor Carbene Complex* [Science](#), **342**, 351-354 (2013)

Survival of the weakest, catalyst design by bonding analysis



J. Wassenaar, E. Jansen, W.-J. van Zeist, F. M. Bickelhaupt, M. A. Siegler, A. L. Spek, and J. N. H. Reek, *Catalyst selection based on intermediate stability measured by mass spectrometry*. [Nature Chem.](#) **2**, 417 (2010)